

HIGH PRESSURE STUDIES OF URANIUM INTERMETALLICS - HYDROGEN SYSTEMS

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Intermetallic compounds with a general formula UTX ($T = 3d, 4d$, or $5d$ atom, and $X = p$ -electron atom like Si, Ge, Al, In) are among widely studied systems. One of the most prominent structure types in which these compounds crystallize is the hexagonal ZrNiAl-type structure (space group $P\bar{6}2m$) [1]. The interest in this family of compounds stems mainly from their unusual magnetic properties [2].

Uranium and its compounds are used in nuclear industry, but several non-nuclear applications have been proposed [3, 4]. One of the possible applications is a hydrogen storage material. It is well known that uranium metal can absorb hydrogen at room temperature. The uranium intermetallic compounds with Laves phase structure type such as UNi₂ and UAl₂ showed an inertness towards hydrogen absorption up to 2 to 10 MPa [5]. However, large hydrogen capacities were reported for the UAl (T=Mn, Fe, Ni) with the hexagonal Fe₂P-type structure (space group $P\bar{6}2m$) [5, 6]. For example, UNiAl form two hydride/deuteride phases UNiAlH_x ($x = 0.7 - 0.8$ for first one and $x = 2.0 - 2.7$ for the second).

In this work the hydrogen absorption of series of UTX (UCoAl, UPtAl and UCoSn) compounds (hexagonal ZrNiAl-type structure, space group $P\bar{6}2m$) by using high hydrogen pressure technique has been investigated.

Exposition of UCoAl at 0.5 GPa of hydrogen pressure and 100°C resulted in formation of UCoAlH₄ hydride. Hydrogen pressures lower than 0.1 GPa were insufficient for hydride formation in this alloy. The crystal structure of this new hydride was refined in the hexagonal AlB₂-type structure ($P6/mmm$ space group, $a = 4.1773$ Å and $c = 4.1254$ Å). Magnetic properties of the UCoAl hydride, studied in the temperature range 2 – 300 K and fields up to 6 T, exhibit the

Curie-Weiss susceptibility with effective moment $2.0\mu_B/U$ and paramagnetic Curie temperature - 6 K.

In the case of UPtAl and UCoSn compounds we could not detect absorbed hydrogen after charging at 1 GPa(H₂) and 100°C. However, the crystal structure transformation from the hexagonal ZrNiAl-type (space group $P\bar{6}2m$) to the hexagonal AlB₂-type structure (space group $P6/mmm$) (like in UCoAlH₄ hydride) was observed for UPtAl intermetallic. In the UCoSn system high hydrogen pressure treatment does not change symmetry of its hexagonal structure but is accompanied with a small change of lattice parameters (from $a = 7.153$ Å, $c = 4.001$ Å for parent compounds to $a = 7.218$ Å, $c = 3.992$ Å).

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