

# EXAFS STUDY OF $\text{YMn}_2\text{D}_6$ SYNTHESIZED UNDER HIGH PRESSURE DEUTERIUM

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## Introduction

$\text{YMn}_2$  compounds have been studied in detail by various groups as it possesses unusual magnetic and electrical properties. These compounds are known to crystallize in  $C15$  cubic structure ( $Fd-3m$ ) and can form hydrides or deuterides under high pressure of hydrogen or deuterium which results in transformation of structure and modification of physical properties. Synthesis of single phase  $\text{YMn}_2\text{D}_x$  (upto  $x = 4.5$ ) along with structural transformation studies based on D content has been extensively studied by XRD (X-Ray diffraction), NPD (neutron powder diffraction) and EXAFS (extended X-Ray absorption fine structure) [1–3]. It is well known that magnetic properties of  $\text{YMn}_2$  are very sensitive to Mn-Mn distances. Formation of deuteride by absorption of deuterium causes increase of cell parameters and results in structural modification which further influences magnetic properties of basic  $\text{YMn}_2$ .

We have reported the synthesis of  $\text{YMn}_2\text{D}_6$  with high deuterium pressure and structural study has been performed both by XRD and NPD [4,5]. XRD structure refinement results show that the structure of the compound is cubic and space group is ( $F-43m$ ). In which it was assumed that deuterium atoms occupying interstitial sites of  $\text{AB}_3$ ,  $\text{A}_2\text{B}_2$  and  $\text{B}_4$  of tetrahedral structure in which experimental results show that  $\text{A}_2\text{B}_2$  sites are most favorable sites and should be filled first before the other two sites filled up. According to NPD studies it is proved that space group of  $\text{YMn}_2\text{D}_6$  is  $Fm-3m$  which is a super group of  $F-43m$  and better refinement results obtained.

In the present investigation we report the Mn K-edge EXAFS results of  $\text{YMn}_2\text{D}_6$  which is necessary to get more knowledge about the short range order of atoms in  $\text{YMn}_2\text{D}_6$ . Also discussed the bond distances between Mn and Y atoms in the compound which determines the important properties.

## Results and Discussion

The  $\text{YMn}_2\text{D}_6$  sample was synthesized under high pressure of deuterium [3]. EXAFS measurement was carried out using synchrotron radiation with the electron beam energy of 1.5 GeV at the beamline of 17 C of NSRRC (National Synchrotron Radiation Research Center, Taiwan). Measurement was performed at room temperature and the photon energy was calibrated to an accuracy of 0.1 eV according to Mn metal K-edge absorption energies.

The EXAFS spectrum was refined by the program REX 2000. The atomic scattering function  $R(r)$  was calculated as the Fourier transformation of  $\chi(k)$  multiplied by  $k^3$ . Single scattering  $r$  space peaks were selected, a fourier transform was calculated and least-squares fitted with an appropriate model as shown in Fig. 1.

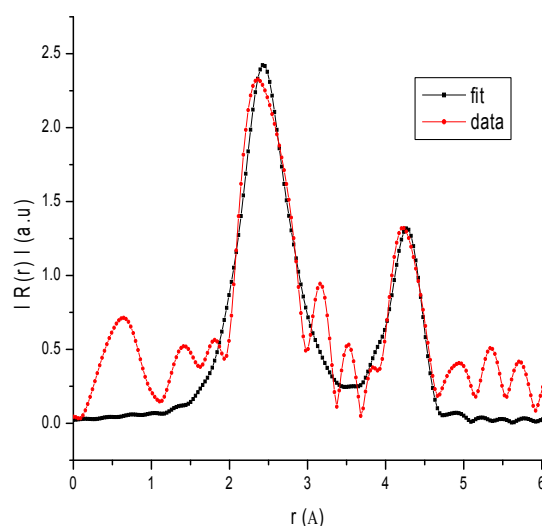


Fig 1. The EXAFS magnitudes  $|R(r)|$  of atomic scattering function of  $\text{YMn}_2\text{D}_6$  at room temperature.

The coordination numbers and interatomic distances from the core atom (Mn2) have been presented in Table 1. Interatomic distances obtained from EXAFS refinement closely resembles NPD results [5] which indicates confirmation of structural pattern positioning atoms in  $\text{YMn}_2\text{D}_6$  based on space group  $Fm-3m$ .

Table 1. EXAFS parameters and their errors obtained from refinement

| Shell   | N (coordination number) | r (Å)        |
|---------|-------------------------|--------------|
| Mn2-Mn1 | 4.038 (0.03)            | 3.057 (0.07) |
| Mn2-Y   | 3.962 (0.03)            | 2.863 (0.01) |
| Mn2-Mn2 | 2.246 (0.02)            | 4.708 (0.04) |

Figure 2 shows the schematic representation of  $\text{YMn}_2\text{D}_6$  based  $Fm-3m$  space group which was derived from NPD data refinement [5]. Figure 1 shows two peaks for three shells which indicate the merging of Mn2-Mn1 peak and Mn2-Y peak because of reduction in bond distance of Mn-Mn and Mn-Y. Due to this shorter bond distance almost similar to  $\text{YMn}_2\text{D}_{4.5}$  (around 2.9 to 3.0 Å), it is expected to have an ordered magnetic property for  $\text{YMn}_2\text{D}_6$  also. But due to the formation octahedral cage of D atoms around Mn2 atoms prevents Mn2-Mn1 interaction and which further restricts long range ordered magnetic interactions.

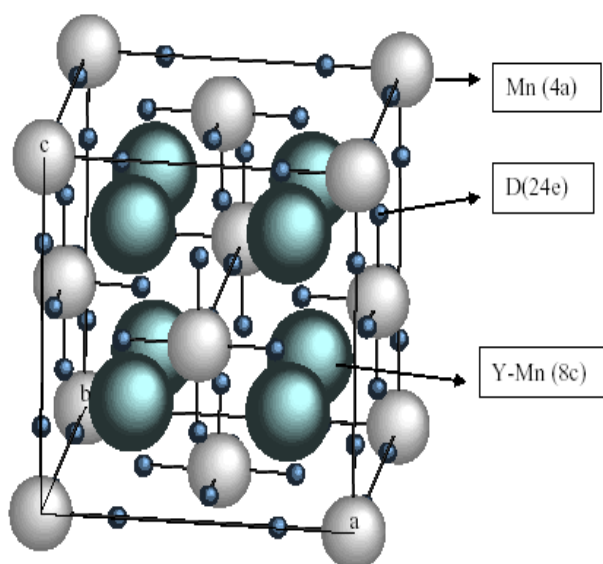


Fig 2. Crystal structure of  $\text{YMn}_2\text{D}_6$  ( $Fm-3m$ )

## Conclusions

The EXAFS measurement of  $\text{YMn}_2\text{D}_6$  has been conducted at Mn K-edge and refinement results were analysed in continuation of our previous synthesis and characterization of  $\text{YMn}_2\text{D}_6$  by XRD and NPD studies. It is observed that bond distances obtained by refinement results are very well comparable with NPD data.

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## References

1. Latroch M, Paul-Bocour V, Przewoznik J, Percheron-Guegan A, Bouree-Vigner F. Neutron diffraction study of  $\text{YMn}_2\text{D}_x$  deuterides ( $1 \leq x \leq 3.4$ ). J Alloys Comp. 1995;231(1-2):99-103.
2. Latroch M, Paul-Bocour V, Percheron-Guegan A, Bouree-Vigner F. Temperature dependence study of  $\text{YMn}_2\text{D}_{4.5}$  by means of neutron powder diffraction. J Alloys Comp. 1998;274(1-2):59-64.
3. Przewoznik J, Paul-Bocour V, Latroch M, Percheron-Guegan A. J Alloys Comp. 1996;232: 107-118.
4. Wang C.Y., Paul-Bocour V., Kang C.C., Liu R.S., Filipek S.M, Dorogova M., Marchuk I., Hirata T., Percheron-Guegan A., Sheu H-S, Jang L.-Y, Chen J.M., Yang H.D., The novel  $\text{YMn}_2\text{D}_6$  deuteride synthesized under high pressure of gaseous deuterium. Solid State Commun. 2004;130:815-820.
5. Paul-Bocour V., Filipek S.M, Dorogova M., Bouree F., Andre G., Marchuk I., Percheron-Guegan A., Liu R.S. Neutron diffraction study, magnetic properties and thermal stability of  $\text{YMn}_2\text{D}_6$  synthesized under high deuterium pressure. J. Solid State Chem. 2005; 178: 356-362.